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Crystal structure of 4-azidomethyl-6-*tert*-butyl-2*H*-chromen-2-oneNasseem El-Khatatneh,<sup>a</sup> Chandra,<sup>a</sup> D. Shamala,<sup>b</sup>  
K. Shivashankar,<sup>b</sup> Mukhokosi Emma Panzi<sup>a</sup> and  
M. Mahendra<sup>a\*</sup><sup>a</sup>Department of Studies in Physics, Manasagangotri, University of Mysore, Mysore 570 006, India, and <sup>b</sup>Department of Chemistry, Central College Campus, Bangalore University, Bangalore 560 001, India. \*Correspondence e-mail: mahendra@physics.uni-mysore.ac.in

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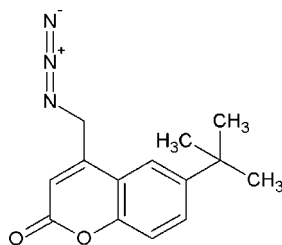
Edited by W. T. A. Harrison, University of Aberdeen, Scotland

In the title compound, C<sub>14</sub>H<sub>15</sub>N<sub>3</sub>O<sub>2</sub>, one of the methyl C atoms of the *tert*-butyl group lies almost in the plane of the chromene ring system [deviation = −0.097 (2) Å], one lies above and one lies below [deviations = 1.460 (3) and 1.006 (3) Å, respectively]. The C—C—N—N torsion angle is 142.33 (17)°. In the crystal, molecules are linked by weak C—H...O hydrogen bonds to generate C(6) chains propagating in the [010] direction.

**Keywords:** crystal structure; chromene; coumarin; hydrogen bonding.**CCDC reference:** 1048730

## 1. Related literature

For background to the biological properties of coumarins, see: Basanagouda *et al.* (2009); Liu *et al.* (2008); Mustafa *et al.* (2011); Ronad *et al.* (2008); Tian *et al.* (2000); Puttaraju *et al.* (2013). For a related structure, see: Chandra *et al.* (2014).



## 2. Experimental

## 2.1. Crystal data

C<sub>14</sub>H<sub>15</sub>N<sub>3</sub>O<sub>2</sub>*M<sub>r</sub>* = 257.29Monoclinic, *P*2<sub>1</sub>/*c*  
*a* = 10.6816 (7) Å  
*b* = 11.1416 (8) Å  
*c* = 11.5409 (8) Å  
 $\beta$  = 100.674 (4)°  
*V* = 1349.72 (16) Å<sup>3</sup>*Z* = 4  
Cu *K*α radiation  
 $\mu$  = 0.71 mm<sup>−1</sup>  
*T* = 293 K  
0.30 × 0.25 × 0.20 mm

## 2.2. Data collection

Bruker X8 Proteum diffractometer  
5911 measured reflections  
2165 independent reflections1949 reflections with *I* > 2σ(*I*)  
*R*<sub>int</sub> = 0.037

## 2.3. Refinement

 $R[F^2 > 2\sigma(F^2)] = 0.045$   
 $wR(F^2) = 0.132$   
*S* = 1.04  
2165 reflections175 parameters  
H-atom parameters constrained  
 $\Delta\rho_{\max} = 0.13 \text{ e \AA}^{-3}$   
 $\Delta\rho_{\min} = -0.17 \text{ e \AA}^{-3}$ 

Table 1

Hydrogen-bond geometry (Å, °).

<i>D</i> —H... <i>A</i>	<i>D</i> —H	H... <i>A</i>	<i>D</i> ... <i>A</i>	<i>D</i> —H... <i>A</i>
C14—H14A...O2 <sup>i</sup>	0.97	2.55	3.311 (2)	135

Symmetry code: (i)  $-x + 1, y - \frac{1}{2}, -z + \frac{1}{2}$ .

Data collection: *APEX2* (Bruker, 2009); cell refinement: *SAINT* (Bruker, 2009); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *PLATON* (Spek, 2009); software used to prepare material for publication: *SHELXL97*.

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Supporting information for this paper is available from the IUCr electronic archives (Reference: HB7363).

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## supporting information

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## Crystal structure of 4-azidomethyl-6-*tert*-butyl-2*H*-chromen-2-one

Nasseem El-Khatatneh, Chandra, D. Shamala, K. Shivashankar, Mukhokosi Emma Panzi and M. Mahendra

### S1. Comment

Coumarin and its substituents are of well known heterocyclic compounds, which have a variety of biologically activities; such as anti-tumour (Mustafa *et al.*, 2011), anti-bacterial (Basanagouda *et al.*, 2009; Liu *et al.*, 2008) and analgesic (Ronad *et al.*, 2008) agents. In addition, coumarin derivatives have been found to be very useful in many applications; such as nonlinear optical materials and as intermediates for the drug synthesis (Tian *et al.*, 2000). In our previous work (Puttaraju *et al.*, 2013), we have reported the synthesis, *in vitro* antimicrobial and anticancer activities of new coumarin derivatives substituted dihydrobenzo[4,5]imidazo[1,2-*a*]pyrimidin-4-ones. In continuation to this, we have synthesized the title compound to study its molecular and crystal structure.

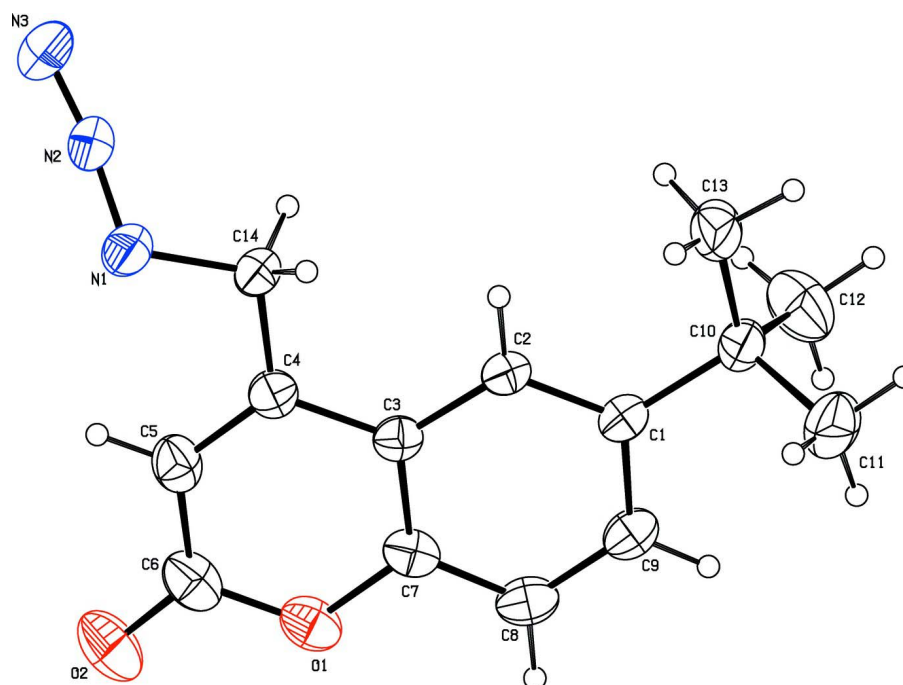
In the molecular structure of the title compound (Fig. 1), the chromene moiety is almost planar, with the maximum deviation from the mean plane being 0.093 (2) Å for atom C10, respectively. The azidomethyl group is in *anti-periplanar* conformation with respect to the chromene moiety, as indicated by the torsion angle value of 172.35 (14)° (C3–C4–C14–N1). The bond lengths and angles are within normal ranges and are comparable to related structure (Chandra *et al.*, 2014). The crystal structure features C—H···O hydrogen bonds, which link the molecules into [010] chains, as shown in Fig. 2.

### S2. Experimental

6-*tert*-Butyl-4-bromomethylcoumarins (0.001 mmol, 0.5 g) were taken in 15 ml acetone in a round bottomed flask and stirred. To this, sodium azide (0.002 mol, 0.13 g) in 5 ml of water was added drop wise with stirring, which was continued for 3 hrs (reaction was monitored by TLC). The reaction mixture was poured in to ice cold water, separated solid was filtered and recrystallized from ethyl alcohol to get pale yellow blocks of the title compound.

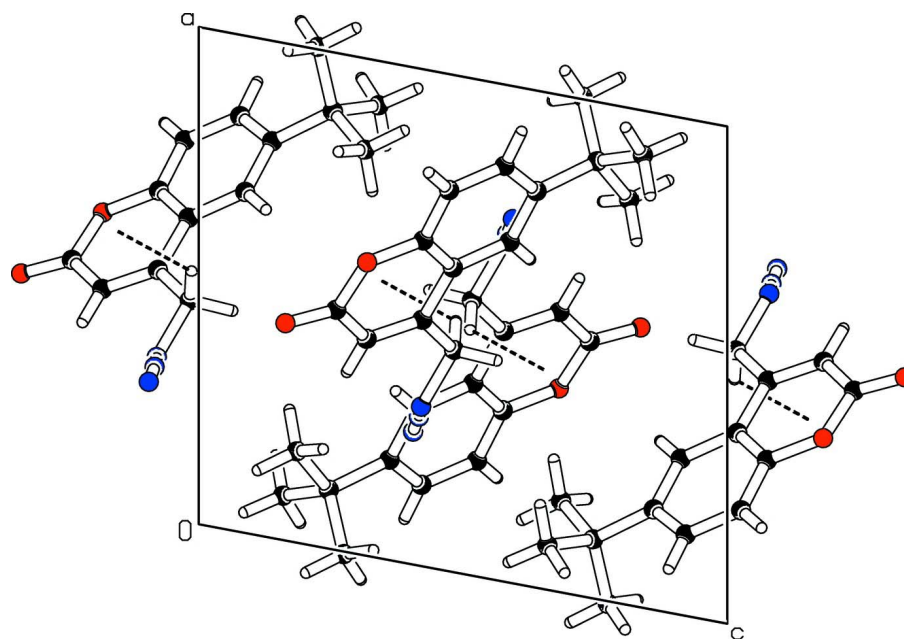
### S3. Refinement

The H atoms were positioned geometrically and allowed to ride on their parent atom, with C–H distance in the range of 0.93 to 0.97 Å;  $U_{\text{iso}}(\text{H}) = 1.2\text{--}1.5U_{\text{eq}}$  (carrier atom) for all H atoms.



**Figure 1**

Perspective diagram of the molecule with 50% probability displacement ellipsoids.



**Figure 2**

Packing diagram of the molecule viewed parallel to the *b* axis.

4-Azidomethyl-6-*tert*-butyl-2*H*-chromen-2-one

## Crystal data

C<sub>14</sub>H<sub>15</sub>N<sub>3</sub>O<sub>2</sub>  
 $M_r = 257.29$   
 Monoclinic,  $P2_1/c$   
 Hall symbol: -P 2ybc  
 $a = 10.6816$  (7) Å  
 $b = 11.1416$  (8) Å  
 $c = 11.5409$  (8) Å  
 $\beta = 100.674$  (4)°  
 $V = 1349.72$  (16) Å<sup>3</sup>  
 $Z = 4$

$F(000) = 544$   
 $D_x = 1.266$  Mg m<sup>-3</sup>  
 Cu  $K\alpha$  radiation,  $\lambda = 1.54178$  Å  
 Cell parameters from 2165 reflections  
 $\theta = 5.6$ – $64.5^\circ$   
 $\mu = 0.71$  mm<sup>-1</sup>  
 $T = 293$  K  
 Block, pale yellow  
 $0.30 \times 0.25 \times 0.20$  mm

## Data collection

Bruker X8 Proteum  
 diffractometer  
 Radiation source: Bruker MicroStar microfocus  
 rotating anode  
 Helios multilayer optics monochromator  
 Detector resolution: 10.7 pixels mm<sup>-1</sup>  
 $\varphi$  and  $\omega$  scans  
 5911 measured reflections

2165 independent reflections  
 1949 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.037$   
 $\theta_{\text{max}} = 64.5^\circ$ ,  $\theta_{\text{min}} = 5.6^\circ$   
 $h = -12 \rightarrow 12$   
 $k = -12 \rightarrow 12$   
 $l = -13 \rightarrow 13$

## Refinement

Refinement on  $F^2$   
 Least-squares matrix: full  
 $R[F^2 > 2\sigma(F^2)] = 0.045$   
 $wR(F^2) = 0.132$   
 $S = 1.04$   
 2165 reflections  
 175 parameters  
 0 restraints  
 Primary atom site location: structure-invariant  
 direct methods

Secondary atom site location: difference Fourier  
 map  
 Hydrogen site location: inferred from  
 neighbouring sites  
 H-atom parameters constrained  
 $w = 1/[\sigma^2(F_o^2) + (0.0667P)^2 + 0.2532P]$   
 where  $P = (F_o^2 + 2F_c^2)/3$   
 $(\Delta/\sigma)_{\text{max}} < 0.001$   
 $\Delta\rho_{\text{max}} = 0.13$  e Å<sup>-3</sup>  
 $\Delta\rho_{\text{min}} = -0.17$  e Å<sup>-3</sup>

## Special details

**Geometry.** Bond distances, angles *etc.* have been calculated using the rounded fractional coordinates. All su's are estimated from the variances of the (full) variance-covariance matrix. The cell e.s.d.'s are taken into account in the estimation of distances, angles and torsion angles

**Refinement.** Refinement on  $F^2$  for ALL reflections except those flagged by the user for potential systematic errors. Weighted  $R$ -factors  $wR$  and all goodnesses of fit  $S$  are based on  $F^2$ , conventional  $R$ -factors  $R$  are based on  $F$ , with  $F$  set to zero for negative  $F^2$ . The observed criterion of  $F^2 > \sigma(F^2)$  is used only for calculating  $-R$ -factor-obs *etc.* and is not relevant to the choice of reflections for refinement.  $R$ -factors based on  $F^2$  are statistically about twice as large as those based on  $F$ , and  $R$ -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (Å<sup>2</sup>)

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
O1	0.59056 (12)	0.02266 (11)	0.31873 (10)	0.0633 (4)
O2	0.43714 (15)	0.00517 (14)	0.16423 (12)	0.0886 (6)
N1	0.32064 (15)	-0.31631 (13)	0.42079 (19)	0.0852 (7)
N2	0.29968 (12)	-0.42384 (13)	0.41655 (12)	0.0581 (5)
N3	0.26741 (17)	-0.51960 (15)	0.40562 (16)	0.0784 (7)

C1	0.79791 (14)	−0.09815 (13)	0.63907 (13)	0.0464 (5)
C2	0.68585 (13)	−0.15469 (12)	0.58820 (12)	0.0441 (4)
C3	0.61248 (13)	−0.11775 (12)	0.48071 (12)	0.0429 (4)
C4	0.49582 (14)	−0.17576 (13)	0.42299 (13)	0.0478 (5)
C5	0.43662 (16)	−0.13424 (15)	0.31759 (14)	0.0585 (6)
C6	0.48347 (19)	−0.03397 (17)	0.26019 (15)	0.0642 (6)
C7	0.65570 (15)	−0.01975 (13)	0.42495 (13)	0.0493 (5)
C8	0.76681 (17)	0.03930 (15)	0.47367 (16)	0.0593 (6)
C9	0.83629 (16)	−0.00013 (14)	0.57861 (16)	0.0566 (5)
C10	0.88114 (15)	−0.14115 (14)	0.75389 (14)	0.0553 (5)
C11	0.9105 (3)	−0.0370 (2)	0.8404 (2)	0.0925 (9)
C12	1.0051 (2)	−0.1916 (3)	0.7259 (2)	0.0969 (10)
C13	0.8160 (2)	−0.2375 (2)	0.81437 (18)	0.0840 (8)
C14	0.44697 (15)	−0.27965 (14)	0.48332 (16)	0.0581 (5)
H2	0.65810	−0.21990	0.62690	0.0530*
H5	0.36230	−0.17200	0.28040	0.0700*
H8	0.79420	0.10520	0.43560	0.0710*
H9	0.91130	0.03960	0.61060	0.0680*
H11A	0.83230	−0.00230	0.85410	0.1390*
H11B	0.95900	0.02280	0.80810	0.1390*
H11C	0.95890	−0.06580	0.91360	0.1390*
H12A	1.05700	−0.22160	0.79690	0.1450*
H12B	1.05020	−0.12930	0.69340	0.1450*
H12C	0.98610	−0.25570	0.66980	0.1450*
H13A	0.73560	−0.20790	0.82820	0.1260*
H13B	0.86890	−0.25830	0.88820	0.1260*
H13C	0.80210	−0.30740	0.76490	0.1260*
H14A	0.50560	−0.34650	0.48640	0.0700*
H14B	0.44200	−0.25750	0.56360	0.0700*

Atomic displacement parameters ( $\text{\AA}^2$ )

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{12}$	$U^{13}$	$U^{23}$
O1	0.0776 (8)	0.0686 (8)	0.0471 (7)	0.0085 (6)	0.0204 (6)	0.0127 (5)
O2	0.1016 (11)	0.1144 (12)	0.0476 (8)	0.0252 (9)	0.0082 (7)	0.0190 (7)
N1	0.0611 (9)	0.0489 (9)	0.1292 (15)	−0.0037 (7)	−0.0249 (9)	−0.0015 (8)
N2	0.0521 (8)	0.0577 (9)	0.0598 (9)	−0.0068 (6)	−0.0019 (6)	−0.0048 (6)
N3	0.0802 (11)	0.0653 (11)	0.0841 (12)	−0.0237 (8)	0.0010 (9)	−0.0092 (8)
C1	0.0489 (8)	0.0428 (8)	0.0493 (8)	−0.0053 (6)	0.0141 (6)	−0.0055 (6)
C2	0.0509 (8)	0.0375 (7)	0.0453 (8)	−0.0036 (6)	0.0128 (6)	−0.0017 (6)
C3	0.0502 (8)	0.0387 (7)	0.0419 (8)	0.0032 (6)	0.0143 (6)	−0.0056 (6)
C4	0.0539 (8)	0.0422 (8)	0.0467 (8)	0.0079 (6)	0.0080 (6)	−0.0100 (6)
C5	0.0616 (10)	0.0630 (10)	0.0482 (9)	0.0110 (8)	0.0030 (7)	−0.0106 (7)
C6	0.0770 (12)	0.0753 (11)	0.0424 (9)	0.0209 (9)	0.0166 (8)	0.0016 (8)
C7	0.0617 (9)	0.0488 (8)	0.0422 (8)	0.0082 (7)	0.0223 (7)	0.0029 (6)
C8	0.0679 (10)	0.0516 (9)	0.0653 (11)	−0.0080 (8)	0.0307 (8)	0.0070 (7)
C9	0.0556 (9)	0.0527 (9)	0.0644 (10)	−0.0122 (7)	0.0189 (8)	−0.0018 (7)
C10	0.0535 (9)	0.0556 (9)	0.0541 (9)	−0.0116 (7)	0.0027 (7)	−0.0021 (7)

C11	0.1146 (18)	0.0844 (14)	0.0685 (13)	-0.0236 (13)	-0.0093 (12)	-0.0150 (11)
C12	0.0754 (13)	0.1139 (19)	0.0993 (17)	0.0241 (12)	0.0111 (12)	0.0184 (14)
C13	0.0873 (13)	0.0933 (15)	0.0614 (11)	-0.0298 (11)	-0.0126 (10)	0.0242 (10)
C14	0.0529 (9)	0.0432 (8)	0.0715 (10)	-0.0048 (7)	-0.0059 (7)	-0.0042 (7)

*Geometric parameters (Å, °)*

O1—C6	1.370 (2)	C10—C12	1.527 (3)
O1—C7	1.3763 (19)	C10—C13	1.518 (3)
O2—C6	1.208 (2)	C2—H2	0.9300
N1—N2	1.218 (2)	C5—H5	0.9300
N1—C14	1.466 (2)	C8—H8	0.9300
N2—N3	1.121 (2)	C9—H9	0.9300
C1—C2	1.384 (2)	C11—H11A	0.9600
C1—C9	1.398 (2)	C11—H11B	0.9600
C1—C10	1.530 (2)	C11—H11C	0.9600
C2—C3	1.4008 (19)	C12—H12A	0.9600
C3—C4	1.452 (2)	C12—H12B	0.9600
C3—C7	1.389 (2)	C12—H12C	0.9600
C4—C5	1.345 (2)	C13—H13A	0.9600
C4—C14	1.494 (2)	C13—H13B	0.9600
C5—C6	1.435 (3)	C13—H13C	0.9600
C7—C8	1.382 (2)	C14—H14A	0.9700
C8—C9	1.370 (3)	C14—H14B	0.9700
C10—C11	1.525 (3)		
C6—O1—C7	121.29 (13)	C3—C2—H2	119.00
N2—N1—C14	116.07 (15)	C4—C5—H5	119.00
N1—N2—N3	172.26 (18)	C6—C5—H5	119.00
C2—C1—C9	117.04 (14)	C7—C8—H8	120.00
C2—C1—C10	122.92 (13)	C9—C8—H8	120.00
C9—C1—C10	120.01 (14)	C1—C9—H9	119.00
C1—C2—C3	122.70 (13)	C8—C9—H9	119.00
C2—C3—C4	124.58 (13)	C10—C11—H11A	110.00
C2—C3—C7	117.50 (13)	C10—C11—H11B	110.00
C4—C3—C7	117.91 (13)	C10—C11—H11C	109.00
C3—C4—C5	118.81 (14)	H11A—C11—H11B	109.00
C3—C4—C14	118.39 (13)	H11A—C11—H11C	109.00
C5—C4—C14	122.81 (15)	H11B—C11—H11C	109.00
C4—C5—C6	122.65 (16)	C10—C12—H12A	109.00
O1—C6—O2	116.67 (17)	C10—C12—H12B	110.00
O1—C6—C5	117.49 (15)	C10—C12—H12C	109.00
O2—C6—C5	125.85 (18)	H12A—C12—H12B	109.00
O1—C7—C3	121.73 (14)	H12A—C12—H12C	109.00
O1—C7—C8	116.95 (14)	H12B—C12—H12C	109.00
C3—C7—C8	121.31 (14)	C10—C13—H13A	109.00
C7—C8—C9	119.38 (15)	C10—C13—H13B	110.00
C1—C9—C8	122.06 (16)	C10—C13—H13C	109.00

C1—C10—C11	110.18 (14)	H13A—C13—H13B	109.00
C1—C10—C12	108.64 (14)	H13A—C13—H13C	109.00
C1—C10—C13	112.18 (14)	H13B—C13—H13C	109.00
C11—C10—C12	109.72 (19)	N1—C14—H14A	109.00
C11—C10—C13	107.06 (16)	N1—C14—H14B	109.00
C12—C10—C13	109.04 (17)	C4—C14—H14A	109.00
N1—C14—C4	110.84 (14)	C4—C14—H14B	109.00
C1—C2—H2	119.00	H14A—C14—H14B	108.00
C7—O1—C6—O2	−175.54 (16)	C2—C3—C4—C5	−177.54 (15)
C7—O1—C6—C5	4.2 (2)	C2—C3—C4—C14	2.2 (2)
C6—O1—C7—C3	−3.5 (2)	C7—C3—C4—C14	−178.75 (14)
C6—O1—C7—C8	176.23 (16)	C2—C3—C7—O1	179.59 (13)
N2—N1—C14—C4	142.33 (17)	C2—C3—C7—C8	−0.1 (2)
C10—C1—C2—C3	−177.49 (14)	C4—C3—C7—O1	0.5 (2)
C2—C1—C9—C8	0.1 (2)	C4—C3—C7—C8	−179.21 (15)
C10—C1—C9—C8	177.99 (15)	C7—C3—C4—C5	1.5 (2)
C2—C1—C10—C11	−128.80 (18)	C3—C4—C5—C6	−0.6 (2)
C2—C1—C10—C12	110.99 (19)	C14—C4—C5—C6	179.65 (16)
C2—C1—C10—C13	−9.6 (2)	C5—C4—C14—N1	−7.9 (2)
C9—C1—C2—C3	0.3 (2)	C3—C4—C14—N1	172.35 (14)
C9—C1—C10—C12	−66.8 (2)	C4—C5—C6—O1	−2.2 (3)
C9—C1—C10—C13	172.62 (15)	C4—C5—C6—O2	177.53 (19)
C9—C1—C10—C11	53.5 (2)	C3—C7—C8—C9	0.5 (2)
C1—C2—C3—C4	178.72 (14)	O1—C7—C8—C9	−179.18 (15)
C1—C2—C3—C7	−0.3 (2)	C7—C8—C9—C1	−0.5 (3)

*Hydrogen-bond geometry (Å, °)*

<i>D</i> —H $\cdots$ <i>A</i>	<i>D</i> —H	H $\cdots$ <i>A</i>	<i>D</i> $\cdots$ <i>A</i>	<i>D</i> —H $\cdots$ <i>A</i>
C14—H14A $\cdots$ O2 <sup>i</sup>	0.97	2.55	3.311 (2)	135

Symmetry code: (i)  $-x+1, y-1/2, -z+1/2$ .